

## Communication

# Total Synthesis of the Maduropeptin Chromophore Aglycon

Kazuo Komano, Satoshi Shimamura, Masayuki Inoue, and Masahiro Hirama J. Am. Chem. Soc., 2007, 129 (46), 14184-14186• DOI: 10.1021/ja076671f • Publication Date (Web): 23 October 2007 Downloaded from http://pubs.acs.org on February 13, 2009



# **More About This Article**

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Links to the 6 articles that cite this article, as of the time of this article download
- Access to high resolution figures
- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article

View the Full Text HTML





Published on Web 10/23/2007

### Total Synthesis of the Maduropeptin Chromophore Aglycon

Kazuo Komano,<sup>†</sup> Satoshi Shimamura,<sup>†</sup> Masayuki Inoue,<sup>\*,‡</sup> and Masahiro Hirama<sup>\*,†</sup>

Department of Chemistry, Graduate School of Science, Tohoku University, Sendai 980-8578, Japan, and Graduate School of Pharmaceutical Sciences, The University of Tokyo, Hongo, Bunkyo-ku, Tokyo 113-0033, Japan

Received September 5, 2007; E-mail: inoue@mol.f.u-tokyo.ac.jp; hirama@mail.tains.tohoku.ac.jp

Maduropeptin, isolated from the broth filtrate of *Actinomadura madurae*, consists of a 1:1 complex of an acidic, water-soluble carrier protein and a chromophore (Figure 1).<sup>1</sup> Schroeder et al. reported the structure of the maduropeptin chromophore **2** and proposed that the nine-membered ring diyne **2** was the methanol adduct of a structurally unknown labile chromophore. Isolated **2** exhibits potent antitumor and antibacterial activities by means of a mechanism distinct from that of similar enediyne natural products.<sup>2</sup> The highly strained system **3**, which is generated via  $S_N2'$  reaction of the C14-amide nitrogen of **2**, cycloaromatizes spontaneously to the *p*-benzyne biradical, which can efficiently cleave doublestranded DNA by hydrogen abstraction. Therefore, **2** represents a stable prodrug form of the reactive nine-membered enediyne structure.

The complex, highly unsaturated molecular architecture of 2, which includes the bicyclo[7.3.0]enediyne and the 15-membered ansa-macrolactam with atropisomerism, clearly presents a daunting challenge for chemical synthesis.<sup>3,4</sup> In particular, controlling the stereoselectivity of both the C4,13-Z-olefin and the non-biaryl atropselectivity within the macrocycle necessitated the development and application of new strategies. Here, we report the first total synthesis of the aglycon **1** of the maduropeptin chromophore, thus demonstrating solutions to these unique synthetic problems.<sup>5</sup>

We hypothesized that the synthesis of **1** could be attained from the convergent assembly of three fragments (**4**,<sup>3b</sup> **5**, and **6**) by applying the C8-epoxide opening reaction to construct the C2'aryl ether and Sonogashira coupling to link C1 and C2 (Figure 1). These couplings would be followed by nine-membered ring formation at C6 and subsequent macrolactamization at C9'. The C4,13-Z-olefin was to be constructed in the last stage of the synthesis through the facile SmI<sub>2</sub>-mediated 1,2-elimination reaction recently developed in our laboratory for the synthesis of the C-1027 chromophore core, a structurally related enediyne natural product.<sup>6</sup>

Synthesis of aglycon 2 started with the stepwise condensation of the three fragments (Scheme 1). The enantioselective introduction of the C7'-hydroxy group was attained on a 30 g scale by heteroene reaction between 2-methoxypropene and 7 using Jacobsen's catalyst 8.7 The obtained 9 was subjected to ozonolysis and acid treatment to afford  $\beta$ -alkoxy ester **10** as an enantiomerically pure compound (76% for three steps, 98% ee).3b,4d LiAlH4 reduction of ester 10 to primary alcohol 11, followed by TBS removal using TBAF, generated triol 6. Then, CsF-promoted coupling<sup>3a,b,8</sup> between epoxide 4 and the sterically hindered phenol 6 resulted in formation of the aryl ether of 12, the C9'-primary alcohol of which was selectively protected as its TBDPS ether to produce triol 13. After derivatization of C9,10-diol 13 to acetonide 14, the remaining C7'secondary alcohol and the C6-terminal acetylene were masked with MOM and TMS, respectively, leading to 16. The C1-hydroxy group of 17 was in turn liberated with DDQ from p-methoxybenzyloxym-



Figure 1. Structure of the maduropeptin chromophore and its retrosynthesis.

ethyl (PMBM) ether **16** and oxidized to ketone **18** using Dess– Martin reagent.<sup>9</sup> Lithium enolate formation and subsequent treatment with PhNTf<sub>2</sub> converted ketone **18** into enol triflate **19**, which was coupled with acetylene moiety **5** under Sonogashira conditions to deliver **20**.<sup>10</sup>

The stage was now set for construction of the two most characteristic rings, the nine-membered diyne and the 15-membered ansa-macrolactam. Prior to formation of the diyne ring, the C6-TMS group of **20** was removed using K<sub>2</sub>CO<sub>3</sub> in methanol, then alcohol **21** was transformed to aldehyde **22** by Dess–Martin oxidation. After screening various reaction conditions, it was found that a mixture of LiN(SiMe<sub>2</sub>Ph)<sub>2</sub><sup>11</sup> (9 equiv) and CeCl<sub>3</sub> (10 equiv) in THF smoothly promoted the acetylide–aldehyde condensation to furnish diyne **23** with the C5- $\alpha$ -hydroxy group in a completely stereoselective fashion.<sup>12,13</sup> The desired C5 stereochemistry presumably originated from the  $\alpha$ -oriented conformation of the aldehyde fixed by the Ce<sup>3+</sup> five-membered chelate; after derivatization of the obtained **23** to methyl ether **24** (82% yield for two steps), the structure of this intermediate was determined by the NOE data (Figure 2).

For the next lactamization, a series of standard functional group manipulations were performed. Selective removal of the TBS group from bis-silylated **24** using the controlled TBAF treatment and subsequent Mitsunobu-type displacement of C14-OH with N<sub>3</sub> provided azide **26**.<sup>14</sup> Next, C9'-protected primary alcohol **26** was converted to C9'-active ester **29** through four steps, including the TBAF-promoted deprotection of the TBDPS group, the stepwise oxidation from primary alcohol **27** to carboxylic acid **28**, and pentafluorophenyl (PFP) ester formation. Slow addition of the isolated azido-PFP ester **29** to excess triphenylphosphine in THF–

<sup>&</sup>lt;sup>†</sup> Tohoku University. <sup>‡</sup> The University of Tokyo.



<sup>*a*</sup> Reagents and conditions: (a) 2-methoxypropene, **8** (5 mol %), acetone, BaO, 4 °C; (b) O<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C; then Me<sub>2</sub>S; (c) PPTS, MeOH, 76% (3 steps, 98% ee); (d) LiAlH<sub>4</sub>, Et<sub>2</sub>O, reflux, 91%; (e) TBAF, THF, rt, 86%; (f) **4** (1.0 equiv), CsF, DMF, 80 °C, 74%; (g) TBDPSCI, Et<sub>3</sub>N, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, rt, 88%; (h) *p*-TsOH, 2,2-dimethoxypropane, MeOH, 91%; (i) MOMCI, *i*-Pr<sub>2</sub>NEt, CH<sub>2</sub>Cl<sub>2</sub>, 84%; (j) LiN(TMS)<sub>2</sub>, TMSCI, THF, -78 °C to rt, 82%; (k) DDQ, CH<sub>2</sub>Cl<sub>2</sub>/ pH 7 buffer (10/1), 99%; (l) Dess-Martin periodinane, NaHCO<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 100%; (m) LiN(TMS)<sub>2</sub>, PhNTf<sub>2</sub>, THF, -78 °C to rt, 87%; (n) **5** (1.0 equiv), Pd(PPh<sub>3</sub>)<sub>4</sub> (5 mol %), CuL *i*-Pr<sub>2</sub>NEt, DMF; (o) K<sub>2</sub>CO<sub>3</sub>, MeOH, 83% (2 steps); (p) Dess-Martin periodinane, NaHCO<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 100%; (q) LiN(SiMe<sub>2</sub>Ph)<sub>2</sub> (9 equiv), CeCl<sub>3</sub> (10 equiv), THF (22 mN), -35 to -25 °C; (r) MeI, NaH, THF, 0 °C, 82% (2 steps); (s) TBAF, THF, 0 °C, 55%; (t) PPh<sub>3</sub>, DEAD, (PhO)<sub>2</sub>P(O)N<sub>3</sub>, THF, rt; (u) TBAF, THF, 0 °C, 58% (2 steps); (v) Dess-Martin periodinane, CH<sub>2</sub>Cl<sub>2</sub>, rt; (w) NaClO<sub>2</sub>, NaH<sub>2</sub>PO<sub>4</sub>, *t*-BuOH/2-methyl-2-butene/H<sub>2</sub>O (6/3/2), 0 °C; (x) C<sub>6</sub>F<sub>5</sub>OH, EDC, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, rt, 50% (3 steps); (y) PPh<sub>3</sub>, Et<sub>3</sub>N, THF/H<sub>2</sub>O (30/1, 4 mM), 45 °C, 66%; (z) PPTS, *p*-TSOH, 80 (Classing *a*) *p*-(trifluoromethyl)benzoyl chloride, DMAP, CH<sub>2</sub>Cl<sub>2</sub>O °C; (bb) SmI<sub>2</sub>, THF, -50 °C, 15 min, 38% (**33:34** = 1:4.1 in DMSO-*d*<sub>6</sub>, 3 steps); (cc) 8 N HC/CF<sub>3</sub>CH<sub>2</sub>OH (1/2), 0 °C, 7%.

H<sub>2</sub>O (30:1) at 45 °C resulted in formation of the 15-membered macrolactam through the intermediacy of the corresponding C14primary amine (66% yield),<sup>3a,15</sup> thus realizing the synthesis of framework **30** of the chromophore. In order for the present route to be synthetically practical, it was particularly important that the cyclization reactions of both **22** and **29** could be performed under non-high-dilution conditions (22 and 4 mM, respectively) on a gram scale without decreasing the yield.

The last phase of the synthesis was the introduction of C4,13-Z-olefin through the SmI<sub>2</sub>-promoted facile 1,2-elimination of *p*-(trifluoromethyl)benzoates.<sup>6</sup> To prepare for the key reaction, the *p*-methoxy benzylidene group of **30** was removed in the presence of the MOM group using a 1:1 molar mixture of PPTS/*p*-TsOH in methanol to afford 1,2-diol **31**, which was treated with *p*-(trifluoromethyl)benzoyl chloride and DMAP to yield **32**. Treatment of bisbenzoate **32** with SmI<sub>2</sub> at -50 °C for 15 min delivered the

desired C4,13-Z-olefin as a mixture of atropisomers (33:34 = 2:1,38% for three steps); the corresponding E-olefin was not observed in this reaction. As shown in Figure 2, the observed NOEs unambiguously established the structures of 33 and 34. Remarkably, potentially reactive functionalities such as the propargylic aryl and methyl ethers were unchanged, indicating the broad applicability of the present 1,2-elimination strategy. The stereoselective formation of the C4,13-Z-olefin can be explained by the conformational stability of the C4-organosamarium intermediate, generated by reduction of the C4-benzoate (Scheme 2). The conformation 35B, in which both the benzoate and the macrocyclic ring are oriented upward, suffers from the resulting large steric interaction and/or increased ring strain; consequently, 35B, which leads to the E-olefin, is strongly disfavored in comparison to 35A that produces 33 and 34. Thus, the 15-membered macrocycle appeared to kinetically control the sole formation of the Z-olefin.16



Figure 2. NOE and ROE data of the synthetic intermediates (24, 33, and 34).

Scheme 2. Mechanistic Rationale for Formation of 33 and 34 from 32



It is noteworthy that the ratio of atropisomers 33 and 34 highly depends on polarity of the solvent,<sup>17</sup> and that the chromatographically separated isomers equilibrated at room temperature to provide the same mixture of isomers after several hours.<sup>18</sup> Most intriguing is the fact that the acid-promoted global deprotection of the mixture of 33 and 34 gave rise to aglycon 1 as the sole atropisomer that corresponds to the natural chromophore 2. The above results indicate that the rotational barrier of the 2,6-substituted benzene was unexpectedly low in comparison to the closely related C-1027 chromophore,<sup>6,19</sup> and that the energy difference of the deprotected atropisomers was larger than that of their protected counterparts.

In summary, the total synthesis of the aglycon of the maduropeptin chromophore was accomplished for the first time. The key features of this synthesis are (1) the efficient convergent union of the three fragments (4, 5, and 6), (2) the cerium amide promoted nine-membered divne ring formation, (3) the one-pot macrolactam formation from the azide-PFP ester, and (4) the SmI<sub>2</sub>-mediated

reductive 1,2-elimination for the stereoselective construction of the C4,13-Z-olefin. Further studies on the total synthesis of chromophore 2 and its various synthetic analogues from 1 are currently underway in our laboratory.

Acknowledgment. This work was supported financially by SORST, Japan Science and Technology Agency (JST). A fellowship to K.K. from the Japan Society for the Promotion of Science (JSPS) is gratefully acknowledged.

Note Added after ASAP Publication. After this paper was published ASAP October 23, 2007, the structures of several compounds were modified in Figure 1 and Scheme 1. The corrected version was published ASAP October 25, 2007.

Supporting Information Available: General methods and spectroscopic and analytical data for selected compounds (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

#### References

- (a) Zein, N.; Solomon, W.; Colson, K. L.; Schroeder, D. R. *Biochemistry* **1995**, *34*, 11591. (b) Schroeder, D. R.; Colson, K. L.; Klohr, S. E.; Zein, N.; Langley, D. R.; Lee, M. S.; Matson, J. A.; Doyle, T. W. *J. Am. Chem. Soc.* **1994**, *116*, 9351. (c) Hanada, M.; Ohkuma, H.; Yonemoto, T.; Tomita, K.; Ohbayashi, M.; Kamei, H.; Miyaki, T.; Konishi, M.; Kawaguchi, H.; Forenza, S. J. Antibiot. 1991, 44, 403.
- (2) For a review on chromoprotein antibiotics and other enediyne natural products, see: Xi, Z.; Goldberg, I. H. In *Comprehensive Natural Products Chemistry*; Barton, D. H. R., Nakanishi, K., Eds.; Elsevier: Amsterdam, 1999: Vol 7, p 553.
- (3) For synthetic studies of the maduropeptin chromophore from this laboratory, see: (a) Kato, N.; Shimamura, S.; Kikai, Y.; Hirama, M. Synlett 2004, 2107. (b) Kato, N.; Shimamura, S.; Khan, S.; Takeda, F.; Kikai, Y.; Hirama, M. Tetrahedron 2004, 60, 3161. (c) Khan, S.; Kato, N.; Hirama, M. Synlett 2000, 1494.
- (4) For synthetic studies of the maduropeptin chromophore from other 101 Synthetic status of the manappendix enomposed from order laboratories, see: (a) Dai, W.-M.; Fong, K. C.; Lau, C. W.; Zhou, L.; Hamaguchi, W.; Nishimoto, S. J. Org. Chem. **1999**, 64, 682. (b) Roger, C.; Grierson, D. S. Tetrahedron Lett. **1998**, 39, 27. (c) Suffert, J.; Toussaint, D. Tetrahedron Lett. 1997, 38, 5507. (d) Nicolaou, K. C.; Koide, K.: Xu, J.: Izraelewicz, M. H. Tetrahedron Lett. 1997, 38, 3671. (e) Nicolaou, K. C.; Koide, K. Tetrahedron Lett. 1997, 38, 3667. (f) Magnus, P.; Carter, R.; Davies, M.; Elliott, J.; Pitterna, T. Tetrahedron 1996, 52, 6283
- (5) Total synthesis of the proposed structure of kedarcidin, a closely related enediyne chromophore, was reported. Ren, F.; Hogan, P. C.; Anderson, A. J.; Myers, A. G. J. Am. Chem. Soc. 2007, 129, 5381. Inoue, M.; Ohashi, I.; Kawaguchi, T.; Hirama, M. Submitted for
- publication.
- Ruck, R. T.; Jacobsen, E. N. J. Am. Chem. Soc. 2002, 124, 2882.
- (8)Kawata, S.; Hirama, M. Tetrahedron Lett. 1998, 39, 8707
- Dess, D. B.; Martin, J. C. J. Am. Chem. Soc. 1991, 113, 7277
- Sonogashira, K. In Comprehensive Organic Synthesis; Trost, B. M., (10)Fleming, I., Eds.; Pergamon: London, 1990; Vol 3, p 521. (11)
- Masamune, S.; Ellingboe, J. W.; Choy, W. J. Am. Chem. Soc. 1982, 104, 5526.
- (a) Inoue, M.; Kikuchi, T.; Hirama, M. *Tetrahedron Lett.* 2004, 45, 6439.
  (b) Iida, K.; Hirama, M. J. Am. Chem. Soc. 1994, 116, 10310.
  A reagent mixture of LiN(TMS)<sub>2</sub> and CeCl<sub>3</sub> proved to be less effective (12)
- (13)for cyclization of 22 (approximately 30% yield of 23). Lal, B.; Pramanik, B. N.; Manhas, M. S.; Bose, A. K. *Tetrahedron Lett.*
- (14)1977, 18, 1977.
- (15) For related reactions, see: (a) Kurosawa, W.; Kan, T.; Fukuyama, T. J. Am. Chem. Soc. 2003, 125, 8112. (b) Achatz, O.; Grandl, A.; Wanner, K. T. Eur. J. Org. Chem. 1999, 1967.
- (16) The importance of the macrolactam structure for the Z-selectivity was suggested by separate experiments. For instance, when a similar substrate without the 15-membered macrocyclic structure was treated with SmI2, sole formation of the C4,13-E-olefin was observed. See also ref 3a.
- $33:34 = 2.5:1 (C_6D_6), 2.1:1 (CDCl_3), 1:1.6 (CD_3OD), 1:4.1 (DMSO-d_6).$ (18) The facile atropisomerism of kedarcidin synthetic intermediates has been
- observed: Myers, A. G.; Hurd, A. R.; Hogan, P. C. J. Am. Chem. Soc. 2002, 124, 4583. See also ref 5.
- Inoue, M.; Sasaki, T.; Hatano, S.; Hirama, M. Angew. Chem., Int. Ed. (19)2004, 6, 3833

JA076671F